Structure and anisotropic properties of $BaFe_{2-x}Ni_xAs_2$ (x=0, 1, and 2) single crystals

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The crystal structure, electrical resistivity, magnetic susceptibility, and heat capacity of single crystals of BaFe₂As₂, BaNi₂As₂, and BaFeNiAs₂ are reported. BaFe₂As₂ data indicate the equivalence of C(T), $d(\chi T)/dT$, and $d\rho/dT$ results in determining the antiferromagnetic transition at T_N =132(1)K. BaNi₂As₂ shows a structural phase transition from a high-temperature tetragonal phase to a low-temperature triclinic phase ($P\bar{1}$ symmetry) at T_0 =131 K, with superconducting critical temperature T_c =0.69 K. BaFeNiAs₂ does not show any sign of superconductivity and its properties resemble BaCo₂As₂, a renormalized paramagnetic metal.

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I. INTRODUCTION

The discovery of superconductivity in a variety of compounds with Fe2+ or Ni2+ square lattice sheets has attracted great interest, stimulated in part by the resemblance of new materials to the well-studied cuprates. The parents of ironbased superconductors include ZrCuSiAs-type RFeAsO (R ThCr₂Si₂-type BaFe₂As₂,² =lanthanoid),¹ PbFCl-type LiFeAs,³ and PbO-type FeSe.⁴ Like in the cuprates, the superconductivity in the Fe-based compounds is suggested to be unconventional as the electron-phonon coupling is apparently too weak by itself to account for the observed high critical temperatures.5-7 The Fe-based superconductors are low carrier density metals with both hole and electron Fermi pockets.⁸ The materials are close to magnetism, and spinfluctuation pairing scenarios have been suggested.^{8–12} The suppression of magnetism in the parent compounds is found to give way to superconductivity, although their possible coexistence in some doping ranges has been suggested. 9,13-15 The Ni-based superconductors have much lower critical temperatures and include those of LaNiAsO, 16,17 SrNi₂As₂,18 and BaNi₂As₂. ^{19,20} LaNiAsO exhibits bulk superconductivity with $T_c \sim 2.75$ K and is theoretically consistent with phonon pairing. ^{17,21} For Pb-flux-grown BaNi₂As₂ crystals, a superconducting critical temperature at $T_c = 0.7$ K was found. BaNi₂As₂ gives a first-order phase transition at T_0 =130 K upon cooling, and at 137 K upon warming.¹⁹ By analogy with AFe_2As_2 (A=Ba, Sr, Ca), ^{22–25} this transition was tentatively identified in Ref. 19 as a structural transition from a high-temperature tetragonal to low-temperature orthorhombic phase, possibly related to a spin-density-wave (SDW) magnetic transition. However, here we find much lower symmetry for the low-temperature structure of BaNi₂As₂.

Due to the existing interest in the properties of ThCr₂Si₂-type Ba T_2 As₂ (T=transition metal) materials, high-quality single-crystalline samples of BaFe_{2-x}Ni_xAs₂ with x=0, 1, and 2, are grown out of excess TAs (T=Fe, Ni). For BaNi₂As₂, single-crystal-temperature structure and a tri-

clinic structure were refined. The anisotropic electrical resistivity and magnetic-susceptibility results are reported for $BaFe_{2-x}Ni_xAs_2$ (x=0, 1, 2). The thermodynamic and transport properties of $BaFe_2As_2$ appear to be different from that of $BaNi_2As_2$. $BaFeNiAs_2$ shows behavioral similarities to $BaCo_2As_2$. The experimental details below are followed by a discussion of the room and low-temperature crystal structures. The thermodynamic and transport properties will then be presented and discussed, sequentially.

II. EXPERIMENTAL DETAILS

In the preparation of crystals, high-purity elements (>99.9%) were used; the source of the elements was Alfa Aesar. Large single crystals of BaFe₂As₂, BaNi₂As₂, and BaFeNiAs2 were grown out of FeAs, NiAs, and FeAs:NiAs binaries, respectively. Self-flux is preferred over other metal solvents such as Sn or Pb, as flux impurities can become incorporated in the crystals; this is recognized to be particularly important issue in BaFe₂As₂.²⁶ FeAs was prepared by placing mixtures of As and Fe pieces in a silica tube. Fe and As pieces were reacted slowly by heating to 700 °C (dwell 6 h), then to 1065 °C (dwell 10 h). NiAs binary was prepared by placing mixtures of As and Ni powder in a silica tube. These were reacted slowly by heating to 300 °C (dwell 3 h), to 600 °C (dwell 20 h), then to 650 °C (dwell 10 h). For BaFe₂As₂ crystals, a ratio of Ba:FeAs=1:5 was heated for 8 h at 1180 °C under partial argon atmosphere. The ampoule was cooled at the rate of 4 °C/hour, followed by decanting of FeAs flux at 1090 °C.²⁷ For the growth of BaNi₂As₂ single crystals, a ratio of Ba:NiAs=1:4 was heated in an alumina crucible for 10 h at 1180 °C under partial atm argon. This reaction was cooled at the rate of 3 °C/hour, followed by decanting of flux at 1025 °C. The BaFeNiAs₂ crystals were prepared in a similar way, but using a near equal molar admixture of FeAs and NiAs. The typical crystal sizes from all batches were $\sim 6 \times 5 \times 0.2$ mm³. The crystals were brittle well-formed plates with the [001] direction perpendicular to the plane of the crystals. The BaNi₂As₂ crystals were found to be highly air sensitive.

Electron probe microanalysis of a cleaved surface of a $BaNi_2As_2$ single crystal was performed on a JEOL JSM-840 scanning electron microscope using an accelerating voltage of 15 kV and a current of 20 nA with an EDAX brand energy-dispersive x-ray spectroscopy (EDS) device. This gave Ba:Ni:As=1:2:2. EDS analyses on the $BaFeNiAs_2$ crystal indicated Ba:Ni:Fe:As ratio of $\sim 1:1:1:2$.

Powder x-ray diffraction data at room temperature were collected using PANalytical X'Pert Pro MPD with an X'celerator position sensitive detector (Cu K_{α} radiation) for BaFe₂As₂, BaNi₂As₂, and BaFeNiAs₂. For BaNi₂As₂ the data were also collected at 105 K using an Anton Parr TTK450 low-temperature stage. The unit-cell parameters were refined from full pattern LaBail fits using the program FULLPROF.²⁸

Single-crystal x-ray diffraction data on BaNi₂As₂ were also collected at 293 K. For room-temperature data collection the crystal was mounted inside a 0.3 mm capillary filled with argon. Further details for this procedure can be found in Ref. 29. For low-temperature data collection, a small single crystal of BaNi₂As₂ was mounted on the top of a thin glass capillary using the Fombin Z15 oil in a stream of nitrogen gas at 173 K. Further details on crystal mounting can be found in Ref. 30. The single crystal was transferred to a Bruker Apex II diffractometer using cryotongues and placed under the stream of nitrogen gas at 100 K. The single-crystal data were collected at 100(1) K, then the temperature of nitrogen gas was raised to 160(1) K and the data were collected again.

dc magnetization was measured as a function of temperature and field using a Quantum Design Magnetic Property Measurement System (MPMS) for BaFe₂As₂, BaNi₂As₂, and BaFeNiAs₂. For a temperature sweep experiment, the sample was cooled to 1.8 K in zero field (zf) and data were collected by warming from 1.8 to 400 K in an applied field of 10 kOe. The magnetic-susceptibility results are presented per mole of formula unit (cm³/mol), χ , along c and ab crystallographic directions. The polycrystalline averages of the susceptibilities can be estimated by $\chi = [2\chi_{ab} + \chi_c]/3$.

Temperature-dependent electrical resistivity was performed on a Quantum Design Physical Property Measurement System (PPMS). The electrical contacts were placed on samples in standard four-probe geometry, using Pt wires and silver epoxy (EPO-TEK H20E). The resistivity was measured in the ab plane (ρ_{ab}) and c direction (ρ_{c}) for BaFe₂As₂, BaNi₂As₂, and BaFeNiAs₂. Specific-heat data, $C_p(T)$, were also obtained using the PPMS via the relaxation method from 200 K to 2 K. For BaNi₂As₂ data were collected down to 0.4 K using a ³He insert.

III. RESULTS

A. Crystal structures

At room temperature, the structures are identified as the tetragonal $ThCr_2Si_2$ (I4/mmm, Z=2), based on x-ray powder-diffraction data. The lattice constants for $BaFe_2As_2$, $BaFeNiAs_2$, and $BaNi_2As_2$ are shown in Fig. 1, along with

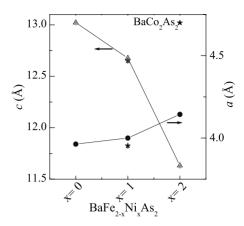


FIG. 1. Room-temperature lattice constants for BaFe₂As₂, BaFeNiAs₂, and BaNi₂As₂, refined from x-ray powder-diffraction data; those for BaCo₂As₂ are shown as stars.

data for BaCo₂As₂. On moving from Fe to Ni, a increases and c decreases. This results in an increased distortion of the tetrahedral environment of the transition-metal atoms (T) in Ba T_2 As₂. The lattice parameters of BaFe₂As₂ are a = 3.9635(5)Å and c = 13.022(2)Å, and those of BaFeNiAs₂ are a = 4.0002(1)Å and c = 12.6767(3)Å, similar to those of BaCo₂As₂. The lattice parameters of BaNi₂As₂ are a = 4.1441(1)Å and c = 11.6325(3)Å, slightly different from those recently reported for the Pb-flux-grown crystals with a = 4.112(4)Å and c = 11.54(2)Å, d and in better agreement with the original reports.

With cooling, $BaFe_2As_2$ and $BaNi_2As_2$ undergo structural phase transitions to lower symmetry. For $BaFe_2As_2$, the transition at 136 K is associated with a tetragonal to orthorhombic *Fmmm* change in space group. Below the phase transition, the four equal Fe-Fe bonds at 280.2 p.m. are split into two pairs with 280.8 p.m. and 287.7 p.m. lengths. For $BaFeNiAs_2$, the low-temperature diffraction was not studied, as no structural transition is expected from the bulk properties (Sec. III B 3). For $BaNi_2As_2$, a tetragonal to orthorhombic first-order phase transition was suggested below T_0 = 130 K only by analogy with AFe_2As_2 (A=Ba, Sr, Ca) (Refs. 22–25) compounds, however, we find that the symmetry of the low-temperature structure is triclinic.

Upon cooling below 130 K, a structural distortion is observed in BaNi₂As₂. Figure 2(a) shows two sections of powder x-ray diffraction data collected at 298 and 105 K, and illustrates the complex diffraction pattern observed at low temperature due to the splitting of the Bragg peaks as the symmetry is reduced. The diffraction pattern at 105 K could only be indexed after removal of all of the symmetry constraints, resulting in a triclinic unit cell. Difficulties related to preferred orientation and the complexity of the low-temperature structure of BaNi₂As₂ precluded satisfactory refinement of the crystal structure from the powder x-ray diffraction data. In order to identify the low-temperature unit cell accurately, single-crystal x-ray diffraction was employed.

The single-crystal x-ray diffraction data for BaNi₂As₂ were collected at 100 K. Due to the symmetry breaking, the low-temperature crystal was twinned, requiring a careful

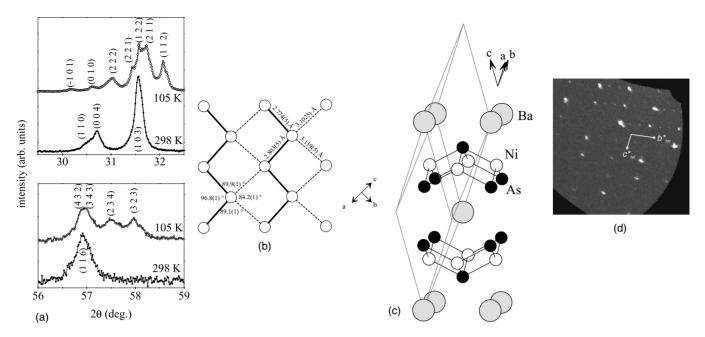


FIG. 2. (a) For $BaNi_2As_2$, two sets of Bragg peaks from powder x-ray diffraction data at 298 and 105 K. (b) The low-temperature triclinic symmetry for $BaNi_2As_2$; the unit cell is represented with lines. (c) The distorted Ni network for low-temperature $BaNi_2As_2$ structure. The Ni atoms form zigzag chains with short and longer Ni-Ni bonds within the plane. (d) The image of the reciprocal space for $BaNi_2As_2$ at 100 K indicates weak superstructure reflections, tripling along the *b* direction.

symmetry analysis before any structural solution and refinement was possible. Because the unit-cell distortion is small and the resolution of the charge-coupled device (CCD) detector is relatively poor, the spots resulting from different twin components could not be spatially resolved for unit-cell determination or intensity integration. Still, indexing of the Bragg reflections suggested a possible monoclinic (α = \sim 89°) or even triclinic symmetry. Because the x-ray powder-diffraction data is immune to twinning problems and provides a larger angular resolution, it was employed to differentiate between the two cells. As described above, the multiple peaks splitting of the powder pattern at 105 K could be indexed only within the triclinic cell.

The structural solution and refinement of the singlecrystal x-ray data were undertaken both in the P1 and P1 space groups. While the P1 space group provided a reasonable solution, strong correlations between different sites containing the same atom types pointed to a higher symmetry. Thus, the structural solution and refinement proceeded in the P1 space group. A numerical absorption correction was based on the crystal shape derived from the optical face indexing. The twinning law accounting for the tetragonal-type twinning (rotation by 90° around the original tetragonal axis) was employed during the final stages of refinement. The refined fraction of the major twin component was 87%, while the scattering contributions of the other three components were 2, 3, and 8%. The structural parameters at 293 and 100 K, from single-crystal x-ray diffraction, are listed in Table I. The resulting triclinic unit cell is outlined in Fig. 2(b), and can be visualized as a distortion of the primitive unit cell of the body-centered ThCr₂Si₂ structure. The powderdiffraction pattern at 105 K was also indexed to a similar triclinic cell [Fig. 2(a)], with a=6.4979(9)Å,

=6.4911(9)Å, c=6.4444(9)Å, α =37.30(1)°, β =54.36(1)°, and γ =37.39(1)°. After heating to 160 K, the single-crystal x-ray data could be indexed in the tetragonal unit cell. The peaks in the 298 K powder-diffraction data were also indexed to the tetragonal cell [Fig. 2(a)].

At room temperature the transition-metal atoms lie on a perfect square net in the ab plane of the tetragonal structure, and for BaNi₂As₂, the Ni-Ni distance is simply $a/2^{1/2}$ = 2.9327(6)Å. In BaNi₂As₂, the further reduction in symmetry below T_0 results in a distorted Ni network as shown in Fig. 2(c). The Ni atoms still lie nearly in a plane, but within the plane form zigzag chains with short Ni-Ni contacts (2.8 Å). The chains are separated by significantly longer Ni-Ni distances (3.1 Å).

Finally, we note that reciprocal space analysis of the 100 K single-crystal x-ray diffraction indicated tripling of one of the directions (the *b* direction within the tetragonal setting), shown in Fig. 2(d). The superstructure reflections were very weak, with their intensities being 2 to 3 times the background level. Because of the low intensities and limited number of the superstructure reflections, no reliable structural solution could be obtained. While the tetragonal-type twinning would require the presence of the superstructure reflections along the other tetragonal direction, the weak nature of the superstructure reflections and dominance of one twin component are the most likely reasons that no superstructure reflections were observed along another direction.

B. Physical properties

1. BaFe₂As₂

For BaFe₂As₂, the magnetic susceptibility at room temperature gives $\chi_c \approx \chi_{ab} \approx 6.7 \times 10^{-4} \text{ cm}^3 \text{ mol}^{-1}$ [Fig. 3(a)].

Temperature (K)		100	293
Crystal system		Triclinic	Tetragonal
Space group		$P\overline{1}$	I4/mmm
Z		1	2
Unit-cell dimensions (Å)		$a=6.5170(14), \ \alpha=37.719(11)^{\circ}$	a=b=4.1474(9)
		$b=6.4587(12), \ \beta=54.009(14)^{\circ}$	c = 11.619(2)
		$c = 6.4440(12), \ \gamma = 37.382(9)^{\circ}$	
Volume (Å ³)		100.13(3)	199.85(7)
2θ range (deg)		3.53°-32.97°	$3.51^{\circ} - 28.62^{\circ}$
Reflections collected		1567	802
Conventional residual (R_1)		0.0682	0.0230
Weighted residual (wR_2) $[I > 2\sigma(I)]$		0.1836	0.0430
Largest diff. peak, hole ($e \text{ Å}^{-3}$)		3.550, -4.076	0.985, -1.365
Atomic coordinates (Wyckoff)		Ba (1 <i>a</i>): 0,0,0	Ba (2a): 0,0,0
		As (2i): 0.3669(3), 0.9785(5), 0.3444(5)	As (4e): 0, 0, 0.3471(1)
		Ni (2i): 0.2219(4), 0.5255(7), 0.7513(7)	Ni(4d): 0, 0.5, 0.25
Isotropic displacement (Ų)	Ba	0.0131(2)	0.0155(3)
	As	0.0277(3)	0.0197(4)
	Ni	0.0366(4)	0.0221(4)

TABLE I. Single-crystal data and structure parameters for $BaNi_2As_2$, at 100 and 293 K. The data were collected at λ =0.71073 Å.

The susceptibility decreases linearly with decreasing temperature, and drops abruptly below ~ 135 K with $\chi_c > \chi_{ab}$ below. As seen in the bottom inset of Fig. 3(a), the field-dependent magnetization is linear at 1.8 K. For this compound, the polycrystalline average of the susceptibility data are presented as $d(\chi T)/dT$ (Ref. 33) versus temperature, to infer an antiferromagnetic transition temperature, at $T_N = 132$ K [the top inset of Fig. 3(a)].

Temperature-dependent electrical resistivity for BaFe₂As₂ decreases with decreasing temperature, consistent with metallic behavior [Fig. 3(b)]. The resistivity behavior is highly anisotropic with $\rho_c > \rho_{ab}$. At room temperature, $\rho_{ab} = 0.50~\text{m}\Omega$ cm and $\rho_c = 16.44~\text{m}\Omega$ cm for BaFe₂As₂. The residual-resistivity ratio (RRR) (= $\rho_{300~\text{K}}/\rho_{2~\text{K}}$) is 4 along both directions. Below ~135 K and for ρ_c and ρ_{ab} there is a sharp steplike drop. This feature is best manifested in the derivative of resistivity, $d\rho/dT$, giving a peak at 132(1) K [bottom inset of Fig. 3(b)]. The magnitude of the resistivity at 2 K and 8 T is higher than the zero-field value for BaFe₂As₂ [top inset of Fig. 3(b)]. This ρ_{ab} rise corresponds to a positive magnetoresistance, $(\rho_{8~\text{T}}-\rho_0)/\rho_0$, of 27.7 % at 2 K. The magnetoresistance for ρ_c is slightly negative and -4.6%, at 2 K.

Figure 3(c) gives the temperature dependence of specific heat. For BaFe₂As₂, specific heat gives a pronounced lambda anomaly, peaking at 132(1) K. Below \sim 6 K, C/T has a linear T^2 dependence [bottom inset of Fig. 3(c)]. The fitted Sommerfeld coefficient, γ , for BaFe₂As₂ is 6.1 mJ/(K² mol) [or 3.0 mJ/(K² mol Fe)]. The Debye temperature is $\theta_D \approx$ 260 K estimated above 150 K, using the calculated values of the T/θ_D dependence of the Debye specific-heat model. Previously, values of γ =16(2)mJ/(K² mol) and θ_D =134(1)K were reported from fits between 3.1 and 14 K for BaFe₂As₂.²

BaFe₂As₂ properties (Fig. 3) demonstrate the consistency among of C(T), $d(\chi T)/dT$ and $d\rho/dT$ in determining T_N = 132(1)K. This transition temperature is lower than the previously reported value of 140 K.²

2. BaNi₂As₂

For BaNi₂As₂, the magnetic susceptibility is anisotropic with $\chi_{ab}/\chi_c \approx 1.6$ at 10 kOe [Fig. 4(a)]. χ is roughly temperature independent and presents only a small drop at $T_0 \approx 132$ K and a rise below ~ 12 K. The field-dependent magnetization is linear at 1.8 K along c and ab crystallographic directions [Fig. 4(a), inset].

BaNi₂As₂ shows metallic behavior [Fig. 4(b)] with ρ_c > ρ_{ab} . At room temperature, ρ_{ab} =0.07 mΩ cm and ρ_c =4.69 mΩ cm. Along c and below T_0 , resistivity abruptly drops two orders of magnitude. Along ab there is a sharp increase at T_0 in ρ followed by a continuous decrease with decreasing temperature. The RRR (= $\rho_{300 \text{ K}}/\rho_{2 \text{ K}}$) is 8 within the ab, suggesting less impurity scattering compared to crystals grown out of Pb.¹⁹ The superconductivity downturn in resistivity starts around 1.5 K in our pure crystal [inset of Fig. 4(b)], similar to the recently reported.¹⁹ T_c as determined by the zero resistance state is 0.69 K, comparable to the crystals grown in Pb flux.¹⁹ There is no noticeable magnetoresistance at 8 Tesla for BaNi₂As₂ at 1.8 K (data not shown).

Figure 4(c) gives the temperature dependence of specific heat of BaNi₂As₂. There is a sharp pronounced transition at T_0 =132 K with a width of less than 0.2 K, in contrast to the broader second-order-like transition observed in BaFe₂As₂. The abrupt changes in both specific heat and ρ_c indicate that the phase transition at T_0 is first-order type. Between 1.8 and ~6 K, the normal-state electronic specific-heat coefficient,

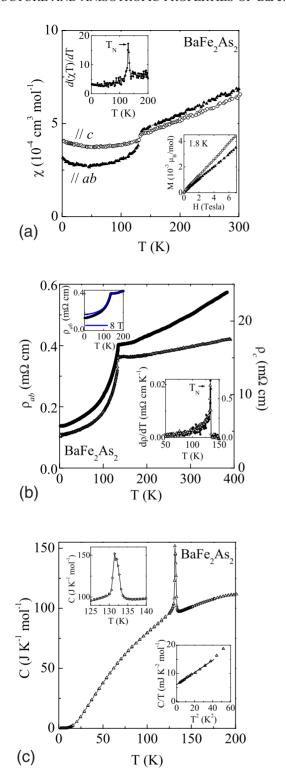


FIG. 3. (Color online) For BaFe₂As₂ and along the two crystal-lographic directions, the temperature dependence of (a) molar susceptibility in zero-field cooled forms measured at 10 kOe, and (b) resistivity at zero field. (c) For BaFe₂As₂, the temperature dependence of the specific heat between 1.8 and 200 K. The inset of (a) shows the field dependence of magnetization and the average susceptibility in the form of $d(\chi T)/dT$ peaking at T_N . Inset of (b) is the derivative in resistivity at zero field, $d\rho/dT$, peaking at T_N ; also $\rho_{ab}(T)$ at 0 and 8 T. Inset of (c) is the C/T vs T^2 and a linear fit between 1.8 K and \sim 7 K.

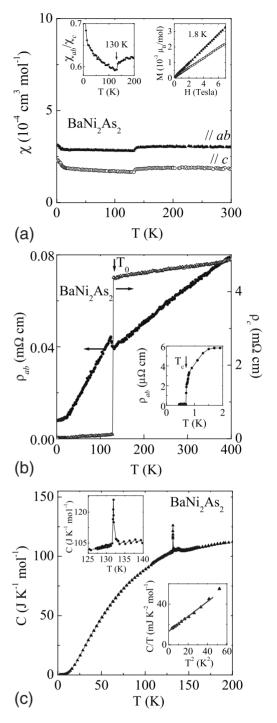


FIG. 4. For BaNi₂As₂ and along the two crystallographic directions, the temperature dependence of (a) molar susceptibility in zero-field cooled forms measured at 10 kOe, and (b) resistivity at zero field. (c) For BaNi₂As₂, the temperature dependence of specific heat between 1.8 and 200 K. The inset of (a) shows the field dependence of magnetization and χ_{ab}/χ_c . Inset of (b) is the plot of $\rho_{ab}(T)$ below 2 K, marking the critical temperature. Inset of (c) is the C/T vs T^2 and a linear fit between 1.8 and \sim 7 K.

 γ_n , is 13.2 mJ/(K² mol) [or 6.6 mJ/(K² mol Ni)]. The Debye temperature is estimated as $\theta_D \approx 250$ K. The value of γ_n is larger than the corresponding value for LaNiAsO_{0.9}F_{0.1}, γ_n =4.75 mJ/(K² mol), ¹⁷ and comparable to other layered

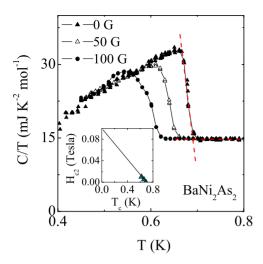


FIG. 5. (Color online) For $BaNi_2As_2$, the field dependence of specific heat below 0.8 K. The T_c was found using the onset criterion demonstrated by the dashed line for the zero-field data. The inset shows the upper critical field H_{c2} for $H \parallel ab$ assigned using the onset criterion, with the solid line representing a least-squares fit to the data.

superconductors Li_xNbS₂ with γ_n =10 mJ/(K² mol) (Ref. 34) and Na_{0.75}CoO₂ γ_n =24 mJ/(K² mol).^{35,36} Previous measurements of Pb-flux-grown crystal gave γ_n =10.8(1)mJ/(K² mol) and θ_D =206 K.¹⁹

For BaNi₂As₂, the temperature dependence of C/T in several fields for $H \parallel ab$ is shown in Fig. 5. In zero field, there is a sharp anomaly at T_c =0.69 K with a jump ΔC =12.6 mJ/(K mol). Taking the value of the Sommerfeld coefficient at T_c , this gives a ratio of $\Delta C / \gamma_n T_c = 1.38$, roughly comparable to a value of 1.43 predicted by weak-coupling Bardeen-Cooper-Schrieffer (BCS) theory and confirms bulk superconductivity. As the magnetic field increases, the transition becomes broader and shifts to lower temperatures. The upper critical field H_{c2} is determined by onset criterion, demonstrated by the dashed line, and displayed in the inset of Fig. 5. The solid line is a linear fit to the data, yielding a slope of $-dH_{c2}/dT_c = -0.142$ T/K. Using the Werthamer-Helfand-Hohenberg $H_{c2}(0) = -0.693 \ T_c (dH_{c2}/dT_c)^{37}$ this gives $H_{c2}(0) = 0.069 \ T$. A zero-temperature, coherence length of $\xi_{GL}(0) \approx 691$ Å is obtained using Ginzburg-Landau (GL) coherence length for- $\xi_{\rm GL} = [\Phi_0/2\pi H_{c2}(0)]^{1/2},$ $\Phi_0 = 2.07$ mula where $\times 10^{-7}$ Oe cm².

3. BaFeNiAs₂

The temperature-dependent magnetic susceptibility at 10 kOe for BaFeNiAs₂ is found to increase with decreasing temperature [Fig. 6(a)]. The averaged polycrystalline magnetic susceptibility at 1.8 K is χ =2.3 × 10⁻³ cm³ mol⁻¹ and at room temperature is 9.3 × 10⁻⁴ cm³ mol⁻¹. These values are comparable to that for BaCo₂As₂.³¹

Temperature-dependent electrical resistivity gives metallic behavior for BaFeNiAs₂ [Fig. 6(b)]. At room temperature, ρ_{ab} =0.08 m Ω cm and ρ_c =2.25 m Ω cm. The RRR is 12 along ab, but only \sim 2 along c. There is no noticeable mag-

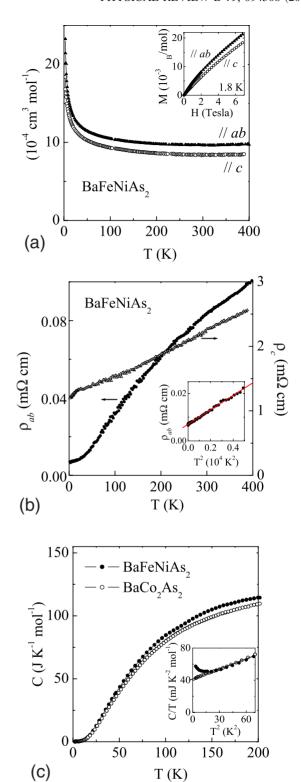


FIG. 6. (Color online) For BaFeNiAs₂ and along the two crystallographic directions, the temperature dependence of (a) molar susceptibility in zero-field cooled forms measured at 10 kOe, and (b) resistivity at zero field. (c) For BaFeNiAs₂ and BaCo₂As₂, the temperature dependence of the specific heat between 1.8 and 200 K. The inset of (a) shows the field dependence of magnetization at 1.8 K. Inset of (b) is the plot of ρ_{ab} vs T^2 and linear fit between 2 and 68 K. The inset of (c) is the C/T vs T^2 and a linear fit between 1.8 and 7 K for BaCo₂As₂.

netoresistance in 8 Tesla (data not shown). For BaFeNiAs₂ the resistivity data exhibit a quadratic temperature dependence, $\rho = \rho_o + AT^2$, over a wide temperature range of 1.8 to $\sim 68~\rm K$. The A value is a measure of the extent of electron correlations. Plotted in the inset of Fig. 6(b) is $\rho(T^2)$, and linear fits in this region. For $\rho_{ab}, A = 3.1 \times 10^{-6}~\rm m\Omega~cm~K^{-2}$ and it is slightly larger than that for BaCo₂As₂ at 2.2 $\times 10^{-6}~\rm m\Omega~cm~K^{-2},^{31}$ consistent with the formation of a Fermi-liquid state.

Figure 6(c) gives the temperature dependence of specificheat results. For BaFeNiAs₂, there are no features in specific heat up to 200 K, similar to that for BaCo₂As₂. The inset of Fig. 5(b) shows C/T versus T^2 dependence. BaFeNiAs₂ gives a slight upturn below ~ 4 K; the γ is remarkably similar to those for BaCo₂As₂. For BaCo₂As₂, C/T vs T^2 is linear below ~ 8 K consistent with a Fermi liquid plus phonon contribution. BaCo₂As₂ gives γ =41.6 mJ/(K² mol) [or 20.8 mJ/(K² mol Co)]. For BaCo₂As₂ the Wilson ratio, $R_w = \pi^2 k_B^2 \chi/(3\mu_B^2 \gamma)$, is ~ 7 from χ_{ab} and ~ 10 from χ_c . ³¹ These values well exceed unity for a free-electron system and indicate that the system is close to ferromagnetism.

IV. FINAL DISCUSSIONS

The shrinking of the crystal lattice is observed from BaFe₂As₂, to BaFeNiAs₂, and finally to BaNi₂As₂. The cell volumes at room temperature are 204.567(2)Å³, 202.8517(3)Å³, and 199.7686(2)Å³, respectively. With cooling, BaFe₂As₂ and BaNi₂As₂ undergo structural transitions. For BaFe₂As₂, a second-order-like transition is observed at 132(1) K, associated with a tetragonal to orthorhombic Fmmm distortion, and a SDW magnetic transition. In contrast to this broader transition, for BaNi₂As₂ there is sharp pronounced feature at T_0 =131 K, seen in $\rho(T)$ and C_p results. This first-order transition was associated with a reduction in the lattice symmetry from tetragonal to triclinic P1. The most interesting effect of the structural distortions in these two compounds is the resulting change in the Fe or Ni network. At room temperature the transition-metal atoms lie on a perfect square net in the ab plane of the tetragonal structure. In BaFe₂As₂, the low-temperature phase is orthorhombic and the Fe network is distorted from square to rectangular, commensurate with magnetic ordering. In BaNi₂As₂, the further reduction in symmetry below T_0 results in a more distorted Ni network, and zigzag chains of Ni atoms. BaFeNiAs₂ shows no evidence of structural or magnetic transitions.

The basic thermodynamic and transport properties of pure $BaFe_2As_2$, $BaNi_2As_2$, and $BaNiFeAs_2$ crystals show differences between them. Unlike $BaFe_2As_2$, the magnetic susceptibility for $BaNi_2As_2$ is roughly temperature independent and presents only a small anomaly at T_0 . At room temperature, χ_{avg} (×10⁻⁴ cm³ mol⁻¹) is smallest for $BaNi_2As_2$ at 3, and slightly larger for $BaFe_2As_2$ at 7 and largest for $BaFeNiAs_2$ at 9. $BaNi_2As_2$ is more conductive at room temperature, compared to $BaFe_2As_2$. The room-temperature resistivity for $BaFeNiAs_2$ and $BaNi_2As_2$ are at ~0.08 m Ω cm, compared to ρ_{ab} =0.50 m Ω cm for $BaFe_2As_2$. At room temperature, ρ_c =16.44 m Ω cm for $BaFe_2As_2$, approximately one order of magnitude larger than for $BaFeNiAs_2$ and $BaNi_2As_2$.

Superconductivity in BaNi₂As₂ is connected to pure sample and those lightly doped with Fe. The BaFe_{2-x}Ni_xAs₂ series of $0 \le x \le 0.2$ gives the highest $T_c^{onset} = 21$ K for x =0.1.38 From temperature dependent resistivity measurements, we find no sign of superconductivity down to 0.4 K for x=0.4, 0.6, 0.8, and 1. The results of BaFeNiAs₂ was given here, as a half point between BaNi₂As₂ (d⁸) and BaFe₂As₂ (d^6). BaFeNiAs₂ is found to show a similar magnitude of magnetic susceptibility as well as specific heat γ to BaCo₂As₂, which was suggested to be in a close proximity to a quantum critical point and of strong fluctuations.³¹ The calculated local-density approximations show hybridization between As p states and Fe d states in BaFe₂As₂ (Ref. 39) similar to BaNi₂As₂ (Ref. 40) and BaCo₂As₂.³¹ However the Fermi level for BaNi₂As₂ is shifted up because of the higher valence electron count in Ni²⁺ (3 d^8) compared to Fe²⁺ (3 d^6). With a larger Fermi surface, BaNi₂As₂ shows large Fermi surfaces and lower density of states. 13,40 Based on bandstructure calculations, the superconductivity of BaNi₂As₂ appears to be rather different from BaFe₂As₂. The Ni phase can be understood within the context of conventional electronphonon theory, while the high- T_c Fe superconductors cannot be described in this way.

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¹Y. Kamihara, T. Watanabe, M. Hirano, and H. Hosono, J. Am. Chem. Soc. **130**, 3296 (2008).

²M. Rotter, M. Tegel, D. Johrendt, I. Schellenberg, W. Hermes, and R. Pottgen, Phys. Rev. B **78**, 020503(R) (2008).

³X. C. Wang, Q. Q. Liu, Y. X. Lv, W. B. Gao, L. X. Yang, R. C. Yu, F. Y. Li, and C. Q. Jin, Solid State Commun. **148**, 538 (2008).

⁴F. C. Hsu, J. Y. Luo, K. W. Yeh, T. K. Chen, T. W. Huang, P. M. Wu, Y. C. Lee, Y. L. Huang, Y. Y. Chu, D. C. Yan, and M. K. Wu, Proc. Natl. Acad. Sci. U.S.A. **105**, 14262 (2008).

⁵L. Boeri, O. V. Dolgov, and A. A. Golubov, Phys. Rev. Lett. 101, 026403 (2008).

⁶I. I. Mazin, D. J. Singh, M. D. Johannes, and M. H. Du, Phys. Rev. Lett. **101**, 057003 (2008).

⁷ A. S. Sefat, M. A. McGuire, B. C. Sales, R. Jin, J. Y. Howe, and D. Mandrus, Phys. Rev. B **77**, 174503 (2008).

⁸R. A. Ewings, T. G. Perring, R. I. Bewley, T. Guidi, M. J. Pitcher, D. R. Parker, S. J. Clarke, and A. T. Boothroyd, Phys. Rev. B **78**, 220501(R), (2008).

⁹C. de la Cruz, Q. Huang, J. W. Lynn, J. Li, W. Ratcliff, J. L.

- Zaretsky, H. A. Mook, G. F. Chen, J. L. Luo, N. L. Wang, and P. Dai, Nature (London) **453**, 899 (2008).
- ¹⁰D. J. Singh and M. H. Du, Phys. Rev. Lett. **100**, 237003 (2008).
- ¹¹ Q. Huang, J. Zhao, J. W. Lynn, G. F. Chen, J. L. Luo, N. L. Wang, and P. Dai, Phys. Rev. B 78, 054529 (2008).
- ¹²I. I. Mazin and M. D. Johannes, Nat. Phys. **5**, 141 (2009).
- ¹³D. J. Singh, M. H. Du, L. Zhang, A. Subedi, and J. An, arXiv:0810.2682 (unpublished).
- ¹⁴H. Chen, Y. Ren, Y. Qiu, Y. Qiu, W. Bao, R. H. Liu, G. Wu, T. Wu, Y. L. Xie, X. F. Wang, Q. Huang, and X. H. Chen, Europhys. Lett. 85, 17006 (2009).
- ¹⁵ A. J. Drew, C. Niedermayer, P. J. Baker, F. L. Pratt, S. J. Blundell, R. H. Lancaster, R. H. Liu, G. Wu, X. H. Chn, I. Watabe, V. K. Malik, A. Dubroka, M. Roessle, K. W. Kim, C. Baines, and C. Bernhard, arXiv:0807.4876, Nature Mater. (to be published).
- ¹⁶T. Watanabe, H. Yanagi, Y. Kamihara, T. Kamiya, M. Hirano, and H. Hosono, J. Solid State Chem. 181, 2117 (2008).
- ¹⁷Z. Li, G. Chen, J. Dong, G. Li, W. Hu, D. Wu, S. Su, P. Zheng, T. Xiang, N. Wang, and J. Luo, Phys. Rev. B **78**, 060504(R) (2008).
- ¹⁸E. D. Bauer, F. Ronning, B. L. Scott, and J. D. Thompson, Phys. Rev. B **78**, 172504 (2008).
- ¹⁹F. Ronning, N. Kurita, E. D. Bauer, B. L. Scott, T. Park, T. Klimczuk, R. Movshovich, and J. D. Thompson, J. Phys.: Condens. Matter 20, 342203 (2008).
- ²⁰N. Kurita, F. Ronning, Y. Tokiwa, E. D. Bauer, A. Subedi, D. J. Singh, J. D. Thompson, and R. Movshovich, arXiv:0811.3426 (unpublished).
- ²¹ A. Subedi, D. J. Singh, M. H. Du, Phys. Rev. B 78, 060506(R) (2008).
- ²²G. Wu, H. Chen, T. Wu, and Y. L. Xie, Y. J. Yan, R. H. Liu, X. F. Wang, J. J. Ying, and X. H. Chen, J. Phys.: Condens. Matter **20**, 422201 (2008).
- ²³C. Krellner, N. Caroca-Canales, A. Jesche, H. Rosner, A. Or-

- meci, and C. Geibel, Phys. Rev. B 78, 100504(R), (2008).
- ²⁴F. Ronning, T. Klimczuk, E. D. Bauer, H. Volz, and J. D. Thompson, J. Phys.: Condens. Matter 20, 322201 (2008).
- ²⁵ N. Ni, S. Nandi, A. Kreyssig, A. I. Goldman, E. D. Mun, S. L. Bud'ko, and P. C. Canfield, Phys. Rev. B 78, 014523 (2008).
- ²⁶ N. Ni, S. L. Bud'ko, A. Kreyssig, S. Nandi, G. E. Rustan, A. I. Goldman, S. Gupta, J. D. Corbett, A. Kracher, and P. C. Canfield, Phys. Rev. B 78, 014507 (2008).
- ²⁷ A. S. Sefat, R. Jin, M. A. McGuire, B. C. Sales, D. J. Singh, and D. Mandrus, Phys. Rev. Lett. **101**, 117004 (2008).
- ²⁸ J. Rodriguez-Carvajal, FULLPROF Suite, Version 3.30, ILL, June 2005.
- ²⁹Y. Mozharivskyj, A. O. Pecharsky, V. K. Pecharsky, and G. J. Miller, J. Am. Chem. Soc. 127, 317 (2005).
- ³⁰M. Gerken, D. A. Dixon, and G. J. Schrobilgen, Inorg. Chem. 39, 4244 (2000).
- ³¹ A. S. Sefat, D. J. Singh, R. Jin, M. A. McGuire, B. C. Sales, and D. Mandrus, Phys. Rev. B **79**, 024512 (2009).
- ³²M. Pfisterer and G. Nagorsen, Z. Naturforsch. C 35b, 703 (1980).
- ³³M. E. Fisher, Philos. Mag. **7**, 1731 (1962).
- ³⁴D. C. Dahn, J. F. Carolan, and R. R. Haering, Phys. Rev. B 33, 5214 (1986).
- ³⁵R. Jin, B. C. Sales, P. Khalifah, and D. Mandrus, Phys. Rev. Lett. 91, 217001 (2003).
- ³⁶ J. L. Luo, N. L. Wang, G. T. Liu, D. Wu, X. N. Jing, F. Hu, and T. Xiang, Phys. Rev. Lett. **93**, 187203 (2004).
- ³⁷N. R. Werthamer, E. Helfand, and P. C. Hohenberg, Phys. Rev. 147, 295 (1966).
- ³⁸L. J. Li, Q. B. Wang, Y. K. Luo, H. Chen, Q. Tao, Y. K. Li, X. Lin, M. He, Z. W. Zhu, G. H. Cao, and Z. A. Xu, arXiv:0809.2009, New J. Phys. (to be published).
- ³⁹D. J. Singh, Phys. Rev. B **78**, 094511 (2008).
- ⁴⁰ A. Subedi and D. J. Singh, Phys. Rev. B **78**, 132511 (2008).