

Phys4125 Formula sheet Exam 1

- Ideal gas law $PV = NkT = nRT$
- Gas pressure from kinetic theory $PV = Nm\overline{v_x^2}$.
- Equipartition

$$\frac{mv_x^2}{2} = \frac{1}{2}kT, \quad U_{thermal} = \frac{f}{2}NkT, \quad f = \text{number of degrees of freedom}$$

- $v_{rms} = \overline{v^2} = \sqrt{\frac{3kT}{m}}$

- Heat and work

$$\Delta U = Q + W, \text{ for quasistatic processes } W = - \int_{V_i}^{V_f} PdV$$

- Adiabatic process $VT^{f/2} = \text{const.}$
- Heat capacities $C = Q/\Delta T$, so that

$$C_V = \left(\frac{\partial U}{\partial T} \right)_{V,N}, \text{ and } C_P = \left(\frac{\partial U}{\partial T} \right)_{P,N} + P \left(\frac{\partial V}{\partial T} \right)_{P,N}$$

- Ideal gas $C_P = C_V + Nk$
- Latent heat $L = Q/m$
- Enthalpy $\Delta H = \Delta U + P\Delta V$.
- Heat conduction

$$\frac{Q}{\Delta t} = -k_t A \frac{dT}{dx}, \quad \frac{\partial T}{\partial t} = K \frac{\partial^2 T}{\partial x^2} \text{ with } K = k_t/c\rho.$$

- Diffusion

$$J_x = -D \frac{\partial n}{\partial x}$$

- Mean free path $l = (4\pi r^2)^{-1}V/N$, time between collisions $\overline{\Delta t} = l/\bar{v}$.
- Transport coefficients of gases

$$k_t = \frac{1}{2} \frac{C_V}{V} l\bar{v}, \quad \eta = \frac{1}{2} \rho \bar{v} l, \quad D = \frac{1}{2} l\bar{v}.$$

- Multiplicity of a coin toss: n heads out of N coins

$$\Omega = \frac{N!}{n!(N-n)!} \equiv \binom{N}{n}$$

- Multiplicity of an Einstein solid: N oscillators, q quanta of energy

$$\Omega = \frac{(q+N-1)!}{q!(N-1)!} \equiv \binom{q+N-1}{q}$$

- Stirling's approximation

$$N! \approx N^N e^{-N} \sqrt{2\pi N} \quad \text{or} \quad \ln N! \approx N \ln N - N.$$

- Multiplicity of a large Einstein solid

$$\ln \Omega \approx (q + N) \ln(q + N) - q \ln q - N \ln N, \quad \Omega \approx \left(\frac{eq}{N}\right)^N, \text{ if } q \gg N.$$

- Multiplicity of a paramagnet

$$\Omega = \frac{N!}{N_{\uparrow}! N_{\downarrow}!} \equiv \binom{N}{N_{\uparrow}}$$

- Surface area of a d -dimensional sphere of radius r

$$Area = \frac{2\pi^{d/2} r^{d-1}}{(d/2 - 1)!}$$

- Multiplicity of an ideal gas in three dimensions

$$\Omega_N \approx \frac{1}{N!} \frac{V^N}{h^{3N}} \frac{\pi^{3N/2}}{(3N/2)!} (2mU)^{3N/2}.$$

- Entropy $S = k \ln \Omega$

- Sackur-Tetrode equation

$$S = kN \left[\ln \left(\frac{V}{N} \left(\frac{4\pi mU}{3Nh^2} \right) \right) + \frac{5}{2} \right]$$

- Temperature

$$\frac{1}{T} = \left(\frac{\partial S}{\partial U} \right)_{V,N}$$

- Measuring entropies

$$dS = \frac{Q}{T} \text{ (const volume)}, \quad \Delta S = \int_{T_i}^{T_f} \frac{C_V}{T} dT.$$

- Spin-1/2 paramagnet in a field

$$M = N\mu \tanh \frac{\mu B}{kT}$$

- Curie's law

$$M \approx \frac{N\mu^2 B}{kT} \text{ for } \mu B \ll kT$$

Physical constants

$$k = 1.38 \times 10^{-23} \text{ J/K}, \quad N_A = 6.02 \times 10^{23}, \quad R = 8.31 \text{ J/K}$$

$$T(^{\circ}\text{C}) = T(^{\circ}\text{K}) - 273.15, \quad 1 \text{ atm} = 1.01 \times 10^5 \text{ Pa}$$